

Dai-Xi Li, Duan-Jun Xu* and
Yuan-Zhi XuDepartment of Chemistry, Zhejiang University,
People's Republic of China

Correspondence e-mail: xudj@mail.hz.zj.cn

Aquapentakis(1*H*-imidazole- κ N³)cobalt(II)
dibromosuccinate

The title compound, $[\text{Co}(\text{C}_3\text{H}_4\text{N}_2)_5(\text{H}_2\text{O})](\text{C}_4\text{H}_2\text{Br}_2\text{O}_4)$, consists of Co^{II} complex cations and dibromosuccinate anions. The Co atom, water O atom and one imidazole N atom of the complex cation occupy a twofold axis. The anion lies on an inversion centre. The crystal packing is reinforced by O—H \cdots O and N—H \cdots O hydrogen bonds, but there are no π – π stacking interactions.

Received 8 November 2004

Accepted 11 November 2004

Online 20 November 2004

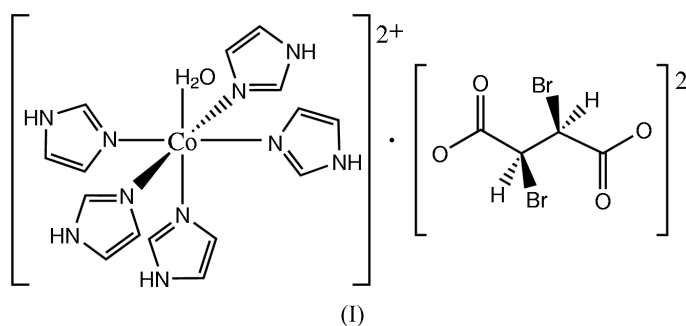
Key indicators

Single-crystal X-ray study
 $T = 295 \text{ K}$
 Mean $\sigma(\text{C}-\text{C}) = 0.007 \text{ \AA}$
 Disorder in main residue
 R factor = 0.037
 wR factor = 0.089
 Data-to-parameter ratio = 13.2

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

Comment

As π – π stacking plays an important role in some biological processes (Deisenhofer & Michel, 1989), the nature of π – π stacking has attracted our attention. A series of metal complexes incorporating different aromatic ligands has been prepared and their structures determined to compare the effect of the size of the aromatic ring on π – π stacking (Liu *et al.*, 2004; Pan & Xu, 2004). The title Co^{II} complex, (I), with imidazole ligands and dibromosuccinate anions, has been prepared but its crystal structure shows no π – π stacking.



The crystal structure of (I) consists of Co^{II} complex cations and dibromosuccinate anions (DBrS), as shown in Fig. 1. The Co^{II} atom, located on a twofold axis, is coordinated by five imidazole ligands and one water molecule in octahedral geometry. The coordinated water O atom and the N5-containing imidazole ligand are also located on the twofold axis. Thus, the N5-imidazole is disordered, the atoms N6 and C8 occupying the same site with 0.5 occupancy each. The N5-imidazole species links with the carboxyl group of DBrS *via* the disordered N6—H6a \cdots O3a or C8—H8 \cdots O3a hydrogen bond (Table 2).

The complete DBrS anion has inversion symmetry. The carboxyl groups display two different orientations (Fig. 1), the dihedral angle between the disordered components of the same carboxyl group being $46(2)^\circ$. The disorder components A (O2a and O3a) and B (O2b and O3b) participate in similar

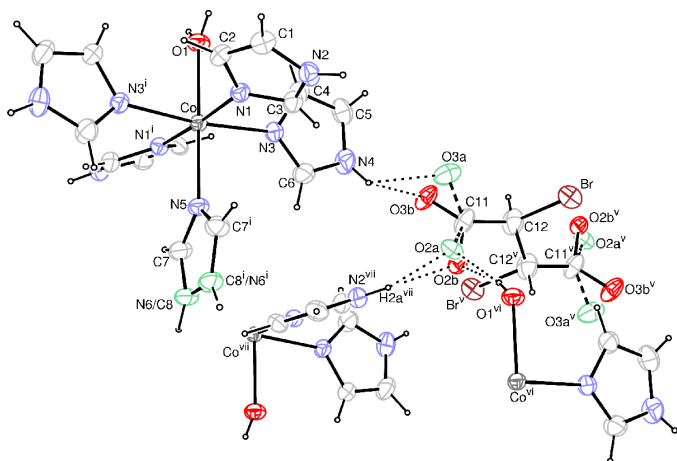


Figure 1

The molecular structure of (I), shown with 30% probability displacement ellipsoids. The atomic label of N6/C8 indicates two different elements occupying the same site within the N5-imidazole ligand. Dashed lines show one of the disordered carboxyl components. Dotted lines indicate hydrogen bonding [symmetry codes: (i) $1 - x, y, \frac{3}{2} - z$; (v) $-x, 1 - y, 2 - z$; (vi) $x - \frac{1}{2}, y - \frac{1}{2}, z$; (vii) $x, 1 - y, z + \frac{1}{2}$].

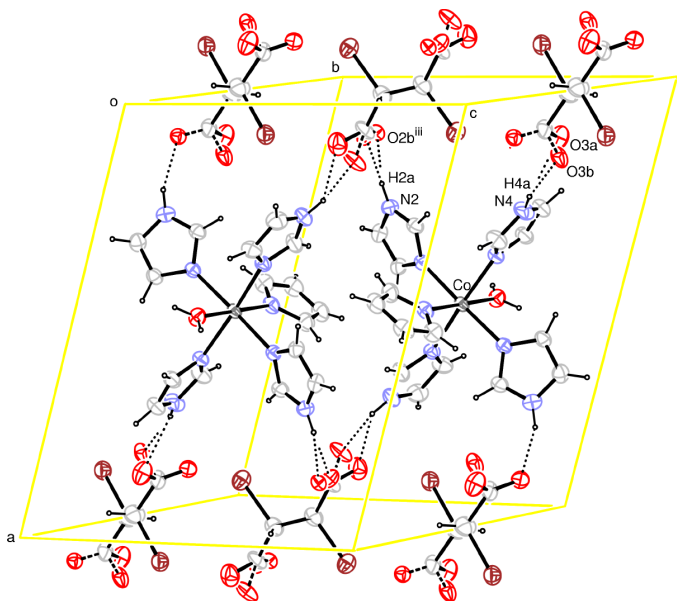


Figure 2

The crystal packing for (I). Dashed lines indicate the hydrogen bonding between imidazole ligands and disordered carboxyl groups [symmetry code: (iii) $x, 1 - y, z - \frac{1}{2}$].

hydrogen-bond linkages with the N2^{vii}-imidazole, N4-imidazole and O1^{vi} water molecule, as shown in Fig. 1 [symmetry code: (vii) $x, 1 - y, z + \frac{1}{2}$]. No π - π stacking is observed in (I). The nearest distance between imidazole rings of neighbouring complex cations is 3.257 (6) Å [N2^{vii}...N2^{viii}; symmetry code: (viii) $x - \frac{1}{2}, \frac{3}{2} - y, z - \frac{1}{2}$] (Fig. 3).

Experimental

An ethanol-water solution (5 ml, 1:1) of CoCl₂·6H₂O (0.24 g, 1 mmol) was mixed with an aqueous solution (4 ml) containing DBrSH₂ (0.55 g, 2 mmol) and Na₂CO₃ (0.21 g, 2 mmol). The mixture

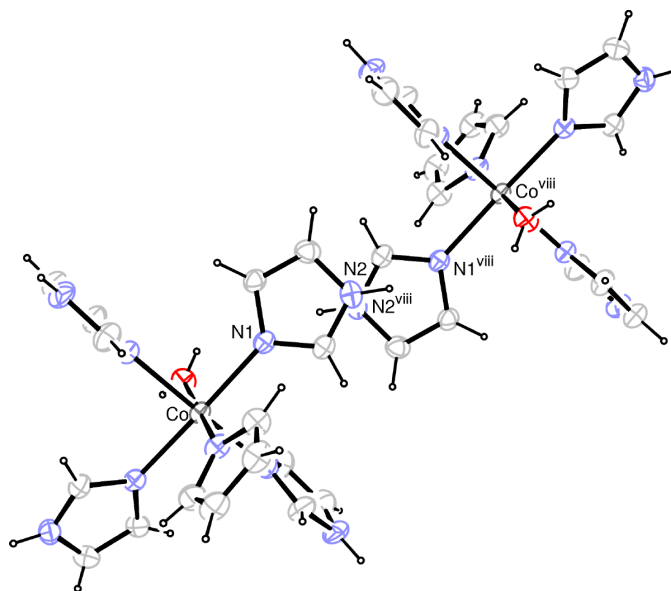


Figure 3

The nearest contact between imidazole rings of neighbouring complex cations [symmetry code: (viii) $x - \frac{1}{2}, \frac{3}{2} - y, z - \frac{1}{2}$].

was refluxed for 4 h and filtered. After cooling to room temperature, imidazole (0.20 g, 3 mmol) was added to the filtrate and pink single crystals of (I) were obtained after 2 d.

Crystal data

[Co(C₃H₄N₂)₅(H₂O)](C₄H₂Br₂O₄)
M_r = 691.23
 Monoclinic, C2/c
a = 15.713 (3) Å
b = 12.798 (2) Å
c = 13.806 (3) Å
 β = 109.468 (9)°
V = 2617.6 (9) Å³
Z = 4

D_x = 1.754 Mg m⁻³
 Mo K α radiation
 Cell parameters from 2322 reflections
 θ = 2.5–25.0°
 μ = 3.76 mm⁻¹
T = 295 (2) K
 Prism, pink
 0.35 × 0.22 × 0.21 mm

Data collection

Rigaku R-Axis RAPID diffractometer
 ω scans
 Absorption correction: multi-scan (ABSCOR; Higashi, 1995)
 T_{\min} = 0.274, T_{\max} = 0.450
 4637 measured reflections

2500 independent reflections
 1880 reflections with $I > 2\sigma(I)$
 R_{int} = 0.017
 θ_{max} = 25.7°
 h = -19 → 19
 k = -15 → 15
 l = -16 → 16

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)]$ = 0.037
 $wR(F^2)$ = 0.089
 S = 1.05
 2500 reflections
 189 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0437P)^2 + 2.2996P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}}$ = 0.002
 $\Delta\rho_{\text{max}}$ = 0.68 e Å⁻³
 $\Delta\rho_{\text{min}}$ = -0.62 e Å⁻³

Table 1

Selected bond lengths (Å).

Co—N1	2.133 (3)	Co—N3	2.164 (3)
Co—O1	2.157 (3)	Co—N5	2.190 (4)

Table 2
Hydrogen-bonding geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O1–H1a \cdots O2a ⁱⁱ	0.85	1.97	2.80 (2)	166
O1–H1a \cdots O2b ⁱⁱ	0.85	1.84	2.67 (2)	166
N2–H2a \cdots O2a ⁱⁱⁱ	0.86	1.93	2.74 (3)	157
N2–H2a \cdots O2b ⁱⁱⁱ	0.86	1.91	2.73 (3)	157
N4–H4a \cdots O3a	0.86	2.23	2.903 (10)	135
N4–H4a \cdots O3b	0.86	1.88	2.670 (11)	152
N6–H6a \cdots O3a ^{iv}	0.85	2.11	2.956 (8)	171
C8–H8 \cdots O3a ^{iv}	0.92	2.05	2.956 (8)	170

Symmetry codes: (ii) $\frac{1}{2} + x, \frac{1}{2} + y, z$; (iii) $x, 1 - y, z - \frac{1}{2}$; (iv) $\frac{1}{2} + x, y - \frac{1}{2}, z$.

Two different orientations for the carboxyl group were found in a difference Fourier map. Their occupancies were initially refined and then fixed as 0.53:0.47 in the final cycles of refinement. H atoms were placed in calculated positions, with C–H = 0.93 (aromatic) and 0.98 Å (methine), and N–H = 0.86 Å. H atoms of water and the disordered imidazole were included with fixed positional parameters and isotropic displacement parameters of 0.08 Å². The other H atoms were refined as riding, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}$ of the carrier atom.

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/MSK, 2002); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* for Windows (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

The project was supported by the National Natural Science Foundation of China (grant No. 20443003 and 29973036).

References

- Altomare, A., Cascarano, G., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). *J. Appl. Cryst.* **27**, 435.
 Deisenhofer, J. & Michel, H. (1989). *EMBO. J.* **8**, 2149–2170.
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
 Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
 Higashi, T. (1995). *ABSCOR*. Rigaku Corporation, Tokyo, Japan.
 Liu, B.-X., Su, J.-R. & Xu, D.-J. (2004). *Acta Cryst.* **C60**, m183–m185.
 Pan, T.-T. & Xu, D.-J. (2004). *Acta Cryst.* **E60**, m56–m58.
 Rigaku (1998). *PROCESS-AUTO*. Rigaku Corporation, Tokyo, Japan.
 Rigaku/MSK (2002). *CrystalStructure*. Version 3.00. Rigaku/MSK, 9009 New Trails Drive, The Woodlands, TX 77381-5209, USA.
 Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.